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Vishal atomic model

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ABSTRACT

This atomic model reveals the new structure of quantum shape of an atom.

Keywords— *Vishal atomic model, the modern atomic model*

1. INTRODUCTION

- (a) Every atom made up of subatomic particles and it can be divisible infinitely.
- (b) Atoms consist of positive and negatively charged energy packets.
- (c) Atom is a set of energy in different mass at different velocities.
- (d) The total system of the atom is always conserved.
- (e) Space between the nucleus and the electron leads to the existence of gravitons and it acts as a medium.
- (f) The atom itself acts as a prism and it is a 3 dimension.
- (g) Electrons fill in the atom according to (EC) theory (electron accommodation theory)

2. ELECTRON ACCOMMODATION THEORY BY VISHAL

- (a) Electrons can accommodate only in the energy level orbitals, which is present inside the shells.
- (b) In a first shell, there are two energy level orbitals each orbital located at two hemispheres, in which each one electron can fill in it.
- (c) In a second shell, there are two energy level orbitals each orbital located at two hemispheres, in which every four electrons can fill in it.
- (d) In a third shell, there are two energy level orbitals each orbital located at two hemispheres, in which every nine electrons can fill in it.
- (e) In a fourth shell, there are two energy level orbitals each orbital located at two hemispheres, in which every sixteen electrons can fill in it.
- (f) If odd no of electrons present in the first, second, third and fourth shell then the electrons mutually shared between the two energy level orbitals.